4 SEM TDC CHMH (CBCS) C 9

2024

(May/June)

CHEMISTRY

(Core)



Paper: C-9

(Organic Chemistry)

Full Marks: 53
Pass Marks: 21

Time: 3 hours

The figures in the margin indicate full marks for the questions

- 1. Choose the correct answer from the following: 1×4=4
 - (a) When acetamide is converted to methanamine, the name of the reaction is
 - (i) Curtius reaction
 - (ii) Michael reaction
 - (iii) Hofmann reaction
 - (iv) Hinsberg reaction

- (b) Which of the following is not an aromatic compound?
 - (i) Furan
 - (ii) Pyrrole
 - (iii) Piperidine
 - (iv) Pyridine
- (c) Which one out of the following is not an alkaloid?
 - (i) Nicotine
 - (ii) Ephedrine
 - (iii) Adrenalin
 - (iv) Quinine
- (d) How many isoprene units are there in diterpene?
 - (i) 1
 - (ii) 2
 - (iii) 3
 - (iv) 4

- 2. Answer any four of the following questions: 2×4=8
 - (a) How will you convert aniline into m-dinitrobenzene? Give the chemical equation.
 - (b) Discuss the basicity of 1°, 2° and 3° amines in aqueous system.
 - (c) Out of pyrrole and furan, which is more aromatic?
 - (d) Thiophene is more aromatic in nature than furan. Explain.
 - (e) What is the difference between terpenes and terpenoids?

UNIT-I

3. Answer any three of the following questions:

3×3=9

(a) Discuss the influence of nitro group upon the basicity of substituted aniline.

- (b) Write short notes on any two of the following: 11/2×2=3
 - (i) Gabriel phthalimide synthesis
 - (ii) Schotten-Baumann reaction
 - (iii) Hinsberg test for 1°, 2° 3° amines
- Complete the following reactions:

1½×2=3

(Continued)

- (i) N-nitroso-N-methyl aniline from N-methylaniline
- (ii) Benzoic acid from aniline
- the following reactions Complete (any three):

(i) $CH_3COCH_2CH_3 + CH_2O + (CH_3)_2NH \longrightarrow$?

(ii)
$$N_2Cl$$
 + OH $pH=9-10$

(iii)
$$\frac{\text{HNO}_3 + \text{H}_2\text{SO}_4}{\text{Low temp.}}$$
?

(iv)
$$\begin{array}{c} \text{NH}_2 \\ + \text{CHCl}_3 & \xrightarrow{\text{KOH}} \end{array}$$
?

UNIT-II

4. Answer any three of the following questions: $3 \times 3 = 9$

Complete following reactions (a) (any three): $1 \times 3 = 3$

(ii)
$$+ CH_2O + HCl \xrightarrow{H_3PO_4}$$
?

Naphthalene

(iii)
$$CrO_3 \rightarrow ?$$

(iv)
$$OH$$
NO₂
NO₂
NO₂
NO₂
Picris acid

Picric acid

Explain why, electrophilic substitution of naphthalene takes place mainly at α-position (C-1).

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- How will you convert any two of the following? 11/2×2=3
 - (i) Naphthalene into Decalin
 - (ii) Anthracene into 9-formylanthracene
 - (iii) Phenanthrene into diphenic acid
- Describe (d) the orientation of sulphonation and Friedel-Crafts acylation in naphthalene. 11/2×2=3

UNIT-III

- Furan, pyrrole and thiophene are less basic than pyridine. Explain.
 - What happens when furan reacted with maleic anhydride under sunlight (hy)? 2
 - Answer any three of the following: $2 \times 3 = 6$
 - (i) Prepare pyrrole from acetylene.
 - (ii) What happens when furfural is treated with acetic anhydride and sodium acetate?

(iii) Complete the following reactions: $1 \times 2 = 2$

(1)
$$H_2 \xrightarrow{\text{Ni}}$$
?

(1)
$$\longrightarrow$$
 + H₂ \longrightarrow ?
(2) \bigcirc + CH₂O + HCl \longrightarrow ?

- $1 \times 2 = 2$ (iv) Convert the following:
 - (1) Pyrrole from furan
 - (2) Pyridine from piperidine
- Pyridine, though aromatic like benzene, can undergo nucleophilic substitution easily, while benzene cannot. Explain.
- What happens when furoic acid is heated up to 200 °C-205 °C?

UNIT-IV

- What are alkaloidal reagents?
 - Explain Hofmann elimination and Emde degradation. How will you (b) differentiate between these two?

Or

How will you convert nicotin into hygrinic acid?

(Turn Over)

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(c)	How will you establish that in nicotine N -methyl pyrrolidine ring is attached to pyridine at position-3 via its α -position?	2
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	Unit—V	
(a)	What is isoprene rule? Explain with suitable example.	1
(b)	Establish the structure of citral. Give its synthesis.	2
	and demand or all the party	
	Discuss the structure of α -Terpineol.	
(c)	How will you synthesize of the following (any one)?	2
	(i) α-Terpineol from α-pinene	
	(ii) Nerol from Neral	

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