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3 TDC (Special) PHY M 4

2016

(July)

#### **PHYSICS**

(Major)

Paper: 30400

## (Condensed Matter Physics and Nuclear Physics)

Full Marks: 67

Time: Three hours

# The figures in the margin indicate full marks for the questions.

Choose the correct option from the following: 1×7=7

- (a) The effective number of atoms in the unit cell of HCP structure is
  - (i) 6
  - (ii) 8
  - (iii) 12
  - (iv) 9

- (b) If K and σ are thermal and electric conductivity of a metal, according Wiedemann-Franz law
  - (i)  $\frac{KT}{\sigma}$  = constant
  - (ii)  $\frac{K\sigma}{T} = \text{constant}$
  - (iii)  $\frac{K}{\sigma T} = \text{constant}$
  - (iv)  $\frac{\sigma}{KT}$  = constant
- (c) The density of charge carriers in a intrinsic semiconductor is proportion
  - (i)  $\exp\left(-\frac{Eg}{KT}\right)$
  - (ii)  $\exp\left(-\frac{2Eg}{KT}\right)$
  - (iii)  $\exp\left(-\frac{Eg}{KT^2}\right)$
  - (iv)  $\exp\left(-\frac{Eg}{2KT}\right)$

- (d) Bohr magnetron  $(\mu_B)$  is
  - (i)  $\frac{e\hbar}{4\pi m_e}$
  - (ii)  $\frac{eh}{4\pi m_e}$
  - (iii)  $\frac{e\hbar}{2\pi m_e}$
  - (iv)  $\frac{eh}{2\pi m_e}$
- (e) Control rods used in nuclear reactors are made up of
  - (i) iron
  - (ii) zinc
  - (iii) cadmium
  - (iv) bismuth
- (f) The experimental electrical method of investigating the size of the nucleus is
  - (i) neutron scattering
  - (ii) α-decay

- (iii) α-particle scattering
- (iv) electron scattering
- of compound nucleus in nuclear fissions
  - (i) surface energy
  - (ii) volume energy
  - (iii) coulomb energy
  - (iv) pairing energy
- 2. (a) Obtain the Miller indices of a plan which intercepts at a,  $\frac{b}{3}$  and 2c in simple cubic unit cell. Draw the diagram showing the plane.
  - (b) What is Hall effect? What important conclusion can be drawn from the Hall 5.
  - (c) Obtain an expression for conductivity of intrinsic semiconductor in terms of mobility.

- 3. (a) What is reciprocal lattice? Derive Bragg's diffraction condition in terms of reciprocal lattice vector. 2+3=5
  - (b) Draw a schematic diagram showing the structure of sodium chloride. Obtain an expression for total cohesive energy of an ionic crystal. What is Madelung constant?

    1+3+1=5
- 4. (a) What do you mean by density of energy states? Show from E-K curve that materials can be classified into conductors, insulators and semiconductors. 3+3=6

### Or

- (b) Discuss briefly the Kronig-Penney model for motion of an electron in a periodic potential. What import conclusions can be drawn from the model?

  3+3=6
- Meissner effect proves the superconductors to be a perfect diamagnetic material. How does the magnetization *M* vary with applied magnetic field *H* in type II superconductors?

  1+3+2=6

- 6. (a) Using semiempirical mass formula, find an expression for energy released is symmetric fission.
  - (b) What are cosmic rays? Discuss briefly about the origin of cosmic rays.

2+2=

- 7. (a) What are the various methods of determining the size of the nucleus Describe any one of them. 2+4=
  - (b) Write the quantum numbers for individual nucleons. Find an expression for electric quadrupole moment.

2+4=0

#### Or

The mass of hydrogen atom and the neutron are 1.008142 a.m.u. and 1.008982 a.m.u. respectively. Calculate the packing fraction of the 16O-nucleus

8. (a) Define nuclear reaction cross-section Describe a method by which the cross section can be determined. What are conserved in nuclear reaction?

(b) Explain the principle and working of a linear accelerator. Show that the length of each drift tube depends on the frequency of r.f. oscillator and gap voltage.

3+3=6

#### Or

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Discuss the method of classification of elementary particles and state their main characteristics. 3+3=6