

**3 SEM TDC CHM M 3 (N/O)**

**2 0 1 7**

( November )

**CHEMISTRY**

( Major )

Course : 303

**( Organic Chemistry—I )**

*The figures in the margin indicate full marks for the questions*

( New Course )

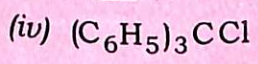
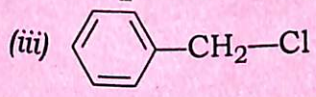
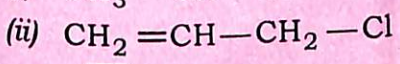
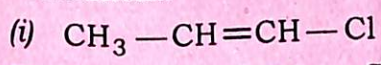
Full Marks : 48

Pass Marks : 14

Time : 2 hours

Select the correct answers from the following : 1×5=5

(a) Which one of the following compounds is most rapidly hydrolyzed by S<sub>N</sub>1 mechanism?





( 2 )

(b) Aldol condensation between which of the following followed by dehydration gives mesityl oxide?

(i) 2 moles of acetaldehyde

(ii) 2 moles of acetone

(iii)  $\text{CH}_3\text{CHO}$  and  $\text{HCHO}$

(iv)  $\text{CH}_3\text{CHO}$  and  $\text{CH}_3\text{COCH}_3$

(c) The product(s) obtained via oxymercuration-demercuration of butyne-1 would be

(i)  $\text{CH}_3-\text{CH}_2-\text{CH}_2-\text{CHO}$

(ii)  $\text{CH}_3-\text{CH}_2\text{CHO} + \text{HCHO}$

(iii)  $\text{CH}_3-\text{CH}_2-\text{CO}-\text{CH}_3$

(iv)  $\text{CH}_3-\text{CH}_2\text{COOH} + \text{HCOOH}$

(d) Among the following compounds, which is most acidic?

(i) *p*-Nitrophenol

(ii) *p*-Hydroxybenzoic acid

(iii) *o*-Hydroxybenzoic acid

(iv) *p*-Toluic acid

(e) The best reagent to convert pent-3-en-2-ol into pent-3-en-2-one is

(i) acidified  $\text{K}_2\text{Cr}_2\text{O}_7$

(ii) alkaline  $\text{KMnO}_4$

(iii) chromic anhydride in glacial acetic acid

(iv) pyridinium chlorochromate

8P/266

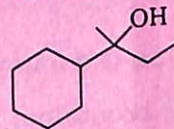
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2. Answer any four of the following questions : 1½×4=6

(a) Giving a suitable example, show that in an  $\text{S}_{\text{N}}2$  reaction inversion takes place.

(b) How would you synthesize the following alcohol from appropriate alkene(s)?



(c) Neo-pentylbromide does not undergo  $\text{S}_{\text{N}}1$  reaction without rearrangement. Explain.

(d) Arrange the following acids in increasing order of acid strength with proper reasoning :

Benzoic acid, salicylic acid,  
*p*-hydroxybenzoic acid and  
*m*-hydroxybenzoic acid

(e) Synthesize fumaric acid from glyoxalic acid and diethyl malonate, using Knoevenagel reaction.

UNIT—I

Answer any two of the following questions : 4×2=8

3. (a) Benzyl chloride can undergo both  $\text{S}_{\text{N}}1$  and  $\text{S}_{\text{N}}2$  reactions with high rate. Explain.

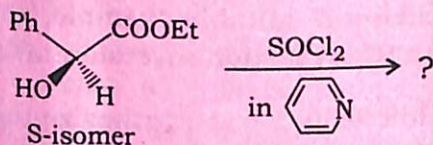
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8P/266



( 4 )

- (b) Complete the following reaction and suggest the mechanism :



4. (a) Comment on the following statement :

In all  $S_N2$  reactions, the rate increases with increasing polarity of the solvent.

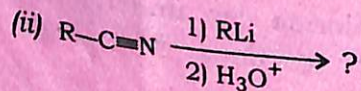
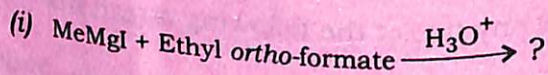
- (b) Synthesize the following :

(i) 1-Bromobutane from pentanoic acid by Hunsdiecker's reaction

(ii) Chloro- or bromo-benzene from aniline by using Sandmeyer reaction

5. (a) Discuss the benzyne mechanism for nucleophilic aromatic substitution reaction. Give evidences in support of the proposed mechanism.

- (b) Complete the following organometallic reactions :

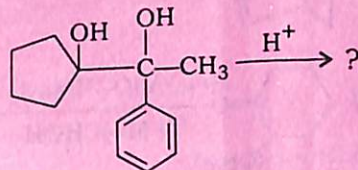


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UNIT—II

2 answer any two of the following questions : 5×2=10

6. (a) Complete the following reaction and suggest the mechanism : 3



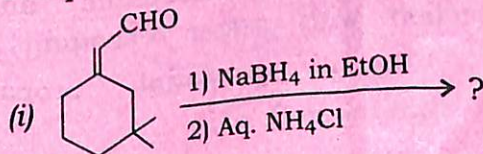
- (b) How is glycerol obtained from propene? Convert glycerol into acrolein. 1+1=2

7. (a) What happens when *p*-cresol is treated with CHCl3 in presence of NaOH? Complete and give the mechanism of the reaction. 3

- (b) Synthesize the following : 1+1=2  
(i) *m*-Nitrophenol from benzene  
(ii) *m*-Cresol from *p*-toluidine

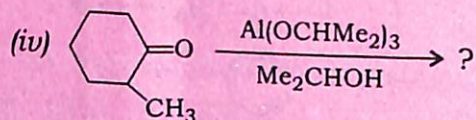
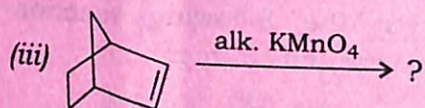
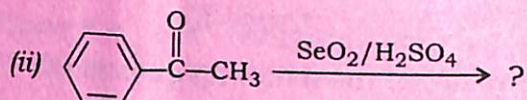
8. (a) Esters of phenols on heating with anhydrous AlCl3 undergo a rearrangement to give phenolic ketones. What is the reaction? Discuss the mechanism involved. 3

- (b) Complete the following reactions :  $\frac{1}{2} \times 4 = 2$





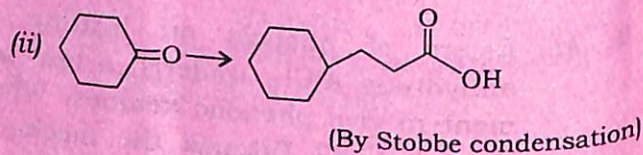
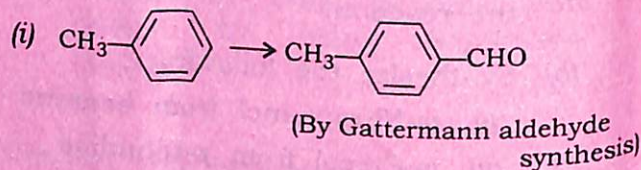
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UNIT—III

Answer any one of the following questions :

9. (a) Carry out synthesis of the following (any one) :



(b) Arrange the following carbonyl compounds in increasing order of their reactivity in nucleophilic addition reaction, with proper reasoning :

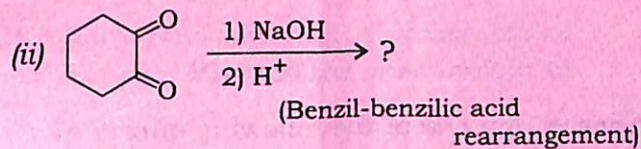
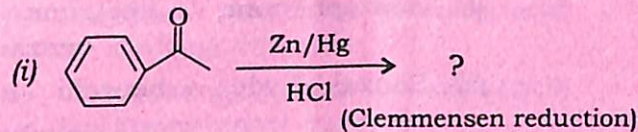
Ethanal, propanal, propanone, butanone-2

8P/266

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( 7 )

(c) Discuss the mechanism of the following reactions : 2×2=4

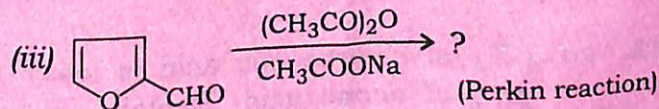
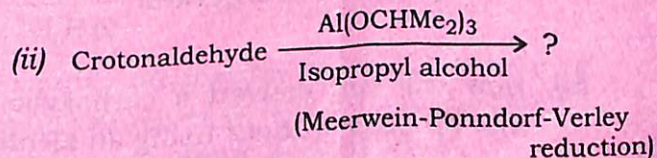
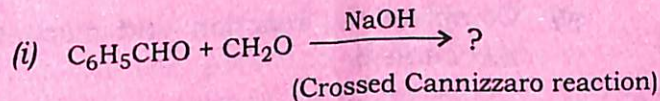


8

10. (a) How will you obtain—

- (i) acetaldehyde from acetylene; 1+1=2  
(ii) benzaldehyde from toluene?

(b) Complete the following reactions and suggest the mechanism (any two) : 2×2=4



2

( Turn Over )



(c) Complete any one of the following conversions : 2

(i) Benzophenone  $\rightarrow$  ketoxime  $\rightarrow$  benzanilide

(ii) Benzaldehyde  $\rightarrow$  benzoin  $\rightarrow$  benzilic acid

#### UNIT—IV

Answer any one of the following questions : 9

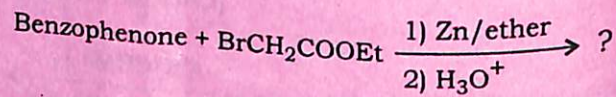
11. (a) Synthesize the following :  $2 \times 2 = 4$

(i) Citric acid from oxalacetic ester by Reformatsky reaction

(ii) Pyruvic acid from acetaldehyde

(b) Convert benzoic acid to phenyl acetic acid by Arndt-Eistert synthesis. 2

(c) Complete the reaction and discuss the mechanism : 2



(d) How will you convert a carboxylic acid into an ester without using an alcohol? 1

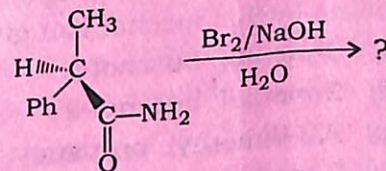
12. (a)  $\text{pK}_a$  of chloroacetic acid is lower than  $\text{pK}_a$  of acetic acid. Explain. 2

(b) Synthesize the following :  $2 \times 2 = 4$

(i) Propanoic acid from butanoic acid by Curtius reaction

(ii) Cinnamic acid from benzaldehyde by Knoevenagel reaction

(c) Complete the following stereochemical reaction and discuss the mechanism of the reaction : 3



#### UNIT—V

Answer any one of the following questions : 2

13. What happens when a thiol reacts with an aldehyde in the presence of HCl? Give a method of preparation of thioether.  $1+1=2$

14. How would you prepare a sulphonic acid by the Strecker reaction? What happens when a thioether is oxidized with  $\text{H}_2\text{O}_2$ ?  $1+1=2$