otal No. of Printed Pages—8

# 6 SEM TDC CHM M 3 (N/O)

2018

(May)

#### CHEMISTRY

(Major)

Course: 603

#### (Inorganic Chemistry—III)

The figures in the margin indicate full marks for the questions

( New Course )

Full Marks: 48

Pass Marks: 14

Time: 2 hours

Choose the correct answer:

 $1 \times 5 = 5$ 

- (a) Non-heme iron protein is
  - (i) hemoglobin
  - (ii) myoglobin
  - (iii) hemerythrin
  - (iv) cytochrome P-450
- (b) The function of plastocyanin is
  - (i) oxidation of L-ascorbic acid
  - (ii) electron transfer in plants
  - (iii) oxidation of primary amine
  - (iv) oxygen transport

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(c)	The	formula of kaolinite clay is
	(i)	Al <sub>2</sub> O <sub>3</sub> ·K <sub>2</sub> SO <sub>4</sub> ·2H <sub>2</sub> O

(ii)  $Al_2O_3 \cdot Na_2SO_4 \cdot 2H_2O$ 

(iii)  $Al_2O_3 \cdot 2SiO_2 \cdot 2H_2O$ 

(iv)  $AI(OH)_3 \cdot CaSO_4 \cdot 2H_2O$ 

Paper chromatography is more suited to

(i) partition

(ii) molecular sieving

(iii) ion exchange

(iv) adsorption

In 1952, the 'Minamata' disease in Japan was caused by poisoning effect of

(ii) Cd

(iii) Hg

(iv) As

## UNIT-I

2. (a) Answer any three questions: 4×3=1

(i) Describe the role of copper in biological system.

(ii) What are the hemoglobin and myoglobin? What functions are the principal similarities in their structures?

(iii) What is carboplatin? Give one of its 3+1=4 uses. What are its advantages over those of cis-platin?

(iv) Explain one function of each of the following metals biological in system:  $2 \times 2 = 4$ 

(1) Molybdenum

(2) Magnesium

Write a note on any one of the following:

(i) Nitrogenase

(ii) Chelation therapy

#### UNIT-II

Answer any three questions:  $3 \times 3 = 9$ 3. (a)

> (i) Discuss about the advantages of solid-state reaction with the help of two examples.

interactions? Give two examples.

(ii) What are the supramolecular

3

3

2

(iii) Mention the two basic approaches for synthesis of nanomaterials. Name the two characterization techniques for nanomaterials.

11/2+11/2=3

(iv) What are clay minerals? Give the formula and uses of montmorillonite clay. 1+2=3

Mention two applications of nanomaterials.

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**4.** (a) Describe the principle and application of paper chromatography.

Or

Define the terms 'stationary phase' and 'mobile phase' in chromatographic process. Name the phases used in TLC.

2+1=

- (b) Write a short note on any one of the following:
  - (i) Principles of gas chromatography
  - (ii) Advantages of TLC over paper chromatography

## UNIT-IV

5. (a) Answer any three questions:

(i) What do you mean by setting of cement? Write down the reactions

(ii) What are paints? Mention the names of essential parts of a paint.
What is the role of a binder?

(iii) What is demineralized water?

Describe a method of deminer-

(iv) Discuss the poisoning effect of mercury (Hg) on human body.

(b) Write short notes on any two of the following: 2×2=4

- (i) Glazing compounds of ceramics
- (ii) Role of thinner in paint industry
- (iii) Hazard from radioactive fallout
- (iv) Composition of cement

#### (Old Course)

Full Marks: 48

Pass Marks: 19

Time: 3 hours

1. Choose the correct answer:

 $1 \times 5 = 5$ 

- (a) Which of the following enzymes do not have heme group?
  - (i) Hemoglobin
  - (ii) Ferrodoxin
  - (iii) Cytochrome oxidase
  - (iv) Catalase
- (b) Which vitamin is known as cyanocobalamin?
  - (i) B<sub>6</sub>
  - (ii) B<sub>12</sub>
  - (iii) K
  - (iv) C

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(Turn Over)